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of $V = V(P, T)$ for a
gas is shown in Fig.
1.1. Determine. the

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two second
derivatives of the
volume of this plot.
(note: the principle
curvatures of the
surface are
proportional to these
second derivatives).

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thermodynamics is

based on

experimental

observations whereas

statistical

thermodynamics is

based on the average

behavior of large

groups of particles.

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Problem 2.8* One
mole of a monatomic
ideal gas at standard
temperature and
pressure (STP)

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undergoes the following three processes: 1. at constant Pressure the Temperature is doubled. 2.

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