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Tournament? First round: Predict
the Gaseous product. Second

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Tournament: Predict the one that dissolves in water to give an acidic solution. Semis: The gas that is more...

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From these historical roots, the study of gases continues to influence our lives. The role of “greenhouse gases,” in particular, remains a vital area of research—and may help determine the winner of the Sweet 16 Chemistry of Gases

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H₂O₂ vs. Yeast (catalyst) 4.)

Na₂SO₃ vs. HCl 5.) Cu vs. HNO₃

(concentrated) 6.) CaC₂ vs. H₂O

7.) NaOCl vs. HCl 8.) NH₄NO₂ vs.

Heat 2nd round- The winner

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dissolves in H_2O to give an acidic solution....

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The compound with a polyatomic ion.

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2.) Zn vs. HCl. 3.) H₂O₂ vs. Yeast
(catalyst) 4.) Na₂SO₃ vs. HCl.

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78% nitrogen. 21% oxygen.
0.03% - 0.04% carbon dioxide.
about 1% argon. small

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Tournament proportions of other noble gases.
varying proportions of water
vapour. previous. 1.

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chemistry - (CCEA) - GCSE ...
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while reviewing these important
topics The rules for filling out the
Tournament brackets are simple:
For the first round,

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