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MAT" HANDBOOK

TRANSPARENCY WORKSHEET

Operations with Scientific Notation I.

Perform the following operations and  
express the answers in scientific notation.

a.  $(1.2 \times 10^5) + (5.35 \times 10^6)$  ('12 (5.35X

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IOC b.  $(6.91 \times 10^{-2}) + (2.4 \times 10^{-3})$

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Scientific Notation 1. Express each of the

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Working not answers  
following numbers in scientific notation. a.  
230 b. 5901 Use with Appendix B,  
Scientific Notation 100000 d. 56 million e.  
2/10 f. 0.4 0 13 g. 0.0 h. 0.000 i. 0.000000  
8 X10<sup>4</sup> 0.000001 j. 12 thousandth l, l K10<sup>4</sup>  
2.

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## Answers MATH HANDBOOK TRANSPARENCY MASTER

### Operations with Scientific Notation

Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal.

Not equal Equal  $(3.4 \times 10^2) + (4.57 \times 10^3)$   
 $= (0.34 \times 10^3) + (4.57 \times 10^3)$  The decimal

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is moved to the left to increase the  
exponent.

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MATH HANDBOOK

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Operations with Scientific Notation I.

Perform the following operations and express the answers in scientific notation.

- a.  $(1.2 \times 10^5) + (5.35 \times 10^6)$  S S b.  $(6.91 \times 10^{-2}) + (2.4 \times 10^{-3})$  c.  $(9.70 \times 10^6) + (8.3 \times$

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105) d.  $(3.67 \times 10^2) - (1.6 \times 10^1)$  e.  $(8.41 \times 10^{-5}) - (7.9 \times 10^{-6})$  f.  $(1.33 \times 10^5) - (4.9 \times 10^4)^2$ .

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Scientific notation takes the form of  $M \times 10^n$  where  $1 < M < 10$  and "n" represents

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the number of decimal places to be moved, Positive  $n$  Indicates the standard form is a large number. Negative  $n$  Indicates a number between zero and one, Example 1: Convert 1,500,000 to scientific notation,

SCIENTIFIC NOTATION Name

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Example 1: Convert 1,500,000 to ...

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Chemistry: Matter and Change 93

Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal.

$(3.4 \times 10^2) + (4.57 \times 10^3) \neq (0.34 \times 10^3) +$

$(4.57 \times 10^3)$  Not equal Equal The decimal

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is moved to the left to increase the  
exponent. # (0.34 " 4.57) ! 103 # 4.91 !  
103 Multiplication

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MASTER 1 ...

Operations With Scientific Notation

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Answers - Displaying top 8 worksheets found for this concept. Some of the worksheets for this concept are Operations scientific notation, Scientific notation mixed operations es1, Writing scientific notation, Writing numbers in scientific notation, Bastien, Name date class 53 math handbook transparency master 1,

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Scientific notation multiplication, Operations  
in scientific notation.

## Operations With Scientific Notation

### Answers Worksheets ...

94 Chemistry: Matter and Change Math  
Handbook Transparency Worksheets 1.  
Perform the following operations and

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express the answers in scientific notation.

a.  $(1.2 \times 10^5) \cdot 6(5.35 \times 10^{-6})$   $(1.2 \times 10^5) (5.35 \times 10^6)$

$(0.12 \times 10^6) (5.35 \times 10^6) (0.12 \cdot 5.35) \cdot 10^6$   $5.47$

$10^6$  b.  $(6.91 \times 10^{-2}) (2.4 \times 10^{-3})$   $(6.91 \times 10^{-2}) (2.4 \times 10^{-3})$   $(6.91 \times 10^{-2}) (0.24 \times 10^{-2})$

Scientific Notation Use with Appendix B.  
Scientific Notation



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96 Chemistry: Matter and Change Math

Handbook Transparency Worksheets 1.

For each of the measurements in the table below, determine if the underlined number is significant or not significant.

Place a check mark in the appropriate box and in the box under the rule that you used to make your determination. 2.

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