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Stoichiometry Model 2 $2A + 3B \rightarrow 5C + 4D$ If 3 mol A react, how many mol C are produced? $3 \text{ mol A} \times \frac{5 \text{ mol C}}{2 \text{ mol A}} = 7.5 \text{ mol C}$ 2 mol A 5. If 21.0 mol A are reacted, how many mol C are produced? Using dimensional analysis, show how you calculated your answer. 6.

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 $2A + 3B \rightarrow 5C + 4D$. 2 mol A produces 5 mol C
4 mol A produces 10 mol C. 3 mol B produces 4 mol D
6 mol B produces 8 mol D

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HS Chemistry POGIL Activity Page 12 Basic

Stoichiometry Model 3 Given the following equation:



25. If 0.052 mol N_2 are reacted, how many mol NH_3 are formed? Using dimensional analysis, show how you calculated your answer.
 $0.052 \text{ mol N}_2 \times 2 \text{ mol NH}_3 = 0.104 \text{ mol NH}_3$
26. 1 mol N_2

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Point out the important parts of their definitions and ...

HS Chemistry POGIL Activity Page 5 Basic

Stoichiometry _____ $\text{H}_2(\text{g}) +$ _____ $\text{O}_2(\text{g})$ _____ $\text{H}_2\text{O}(\text{g})$

14. Given the equation above, determine the number of moles of water produced when 5.2 g O_2 are reacted. Make sure to show the dimensional analysis in your work. [View full document.](#)

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connect to . each other and how do we represent those connections by the names we give a substance? Figure 1 . Particle . model

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HS Chemistry POGIL Activity Name: Date: Basic

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4 mol A produces 10 mol C
3 mol B produces 4 mol D
6 mol B produces 8 mol D ...

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