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Chapter 13 Review Answer Key CHAPTER 13 REVIEW Solutions SECTION 13-2 SHORT ANSWER Answer the following questions in the space provided. 1. Following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid. b.

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ANSWER: For more details: See chapter 13 part 1 video or chapter 13 section 2 in the textbook. 13.2) Draw the line bond structure for the general form of an amino acid using "R" to represent the side chain. Draw the N- and C-terminus groups using the forms that are predominant at physiological pH (pH = 7.4). Include lone pairs in your drawing.

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liquid solution increases the solubility of the gas in the liquid. b.

### CHAPTER 13 REVIEW Solutions

SHORT ANSWER 1. a. Increasing the pressure of the solute gas above the solution puts stress on the equilibrium of the system. Gas molecules collide with the liquid surface more often, causing an increase in the rate of gas molecules entering into solution. b. As the temperature of the water increases, water molecules move faster,

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 2

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. \_\_\_\_\_ a. Predict the boiling point of a 0.200 m solution of glucose in water. \_\_\_\_\_ b. Predict the boiling point of a 0.200 m solution of potassium iodide in water. 2.

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